

435(1): Properties of the Time Dependent Schrödinger Equation in n Space

Consider:
$$E = i\hbar \frac{d\psi}{dt} = H_1 \psi \quad - (1)$$

It follows that:
$$\hbar\omega \int \psi^* \frac{1}{m(r)^{1/2}} \psi d\tau = \langle H_1 \rangle \quad - (2)$$

Denoting:
$$E_1 = \langle H_1 \rangle \quad - (3)$$

it follows that:
$$E_1 = \hbar\omega \int \psi^* \frac{1}{m(r)^{1/2}} \psi d\tau \quad - (4)$$

For example, the Lamb shift in atomic H increases the energy of $2S_{1/2}$, and causes its frequency from ω to ω_1 . The new frequency is:

$$\omega_1 = \omega \int \psi_H^* \frac{1}{m(r)^{1/2}} \psi_H d\tau \quad - (5)$$

so
$$\frac{\omega_1}{\omega} = \int \psi_H^* \frac{1}{m(r)^{1/2}} \psi_H d\tau \quad - (6)$$

Both ω_1 and ω are known with great precision experimentally, so $m(r)^{1/2}$ can be found for the energy change in $2S_{1/2}$. There is no energy change in $2P_{1/2}$ so

in this case:
$$m(r)(2P_{1/2}) = 1 \quad - (7)$$



2) So in this interpretation, each energy level is associated with its own $n(r)^{1/2}$ function. The Lamb shifted energy levels each have $n(r)$ different from one. The unshifted levels all have $n(r)$ of unity. This method removes the necessity of having to model $n(r)$, and each Lamb shift is caused by its characteristic $n(r)$ function.

The energy levels of the Dirac atom are:

$$E_{n,j} = E_n \left(1 + \left(\frac{\alpha}{n} \right)^2 \left(\frac{n}{j+1/2} - \frac{3}{4} \right) \right) \quad (7)$$

where E_n are the energy levels of the Schrodinger atom. Here j is the total angular momentum quantum number:

$$j = L + S, \dots, L - S \quad (8)$$

$$(9)$$

and here α is the fine structure constant. In the Dirac atom there is no Lamb shift because j is the same for $2p_{1/2}$ and $2s_{1/2}$.

$$2s_{1/2} : n=1, L=0, S=1/2, j=1/2$$

$$2p_{1/2} : n=1, L=1, S=-1/2, j=1/2$$

However, it is generally covariant H atom wave function of the Dirac atom is exceedingly complicated, so the ψ_H wave functions are used in the first approximation.