

329(3): Evaluation of the Quantum Relativistic Hamiltonian in the First Approximation

The Hamiltonian is :

$$H_0 = H - mc^2 = \frac{p^2 c^2}{(p^2 c^2 + m^2 c^4)^{1/2} + mc^2} + U \quad (1)$$

i.e:

$$\langle H_0 \rangle = -\frac{e^2 c^2}{p^2} \left\{ \frac{\psi^* \nabla^2 \psi d\tau}{(p^2 c^2 + m^2 c^4)^{1/2} + mc^2} + \int \psi^* U \psi d\tau \right\} \quad (2)$$

In this equation:

$$p^2 = 2m(H_0 - U) \quad (3)$$

in the denominator of the first term on the right hand side of eq. (2). Eq. (3) is the non-relativistic first approximation, in which:

$$U = -\frac{e^2}{4\pi \epsilon_0 r} \quad (4)$$

H_0 is a constant of the classical Hamiltonian motion.

Now we:

$$(p^2 c^2 + m^2 c^4)^{1/2} + mc^2 = (\gamma + 1)mc^2 \quad (5)$$

Take:

$$\gamma = \left(1 - \frac{p_0^2}{mc^2}\right)^{-1/2} \quad (6)$$

Therefore:

$$\langle H_0 \rangle = -\frac{\hbar^2 c^2}{m} \left[\frac{\psi^* \nabla^2 \psi d\tau}{\left(2 + \frac{p_0^2}{2mc^2}\right) mc^2} + \int \psi^* U \psi d\tau \right] \quad (7)$$

where the following approximation is used:

$$\left(1 - \frac{p_0^2}{mc^2}\right)^{-1/2} \approx 1 + \frac{1}{2} \frac{p_0^2}{mc^2} \quad (8)$$

Therefore:

$$\langle H_0 \rangle = -\frac{\hbar^2}{m} \left[\frac{\psi^* \nabla^2 \psi d\tau}{\left(2 + \frac{H_0 - U}{mc^2}\right) m} + \int \psi^* U \psi d\tau \right] \quad (9)$$

$$\overrightarrow{H_0 - U} \left(\frac{1}{mc^2} \right) - \frac{\hbar^2}{2m} \int \psi^* \nabla^2 \psi d\tau + \int \psi^* U \psi d\tau,$$

The non-relativistic result from the Schrödinger equation. This is:

$$H_0 = -\frac{me^4}{32\pi^2 \epsilon_0^2 \hbar^2 n^2} \frac{1}{r} \quad (10)$$

for the energy levels of the hydrogen atom,

where n is the principal quantum number.

More accurately:

$$\langle H_0 \rangle = -\frac{e^2}{m} \int \psi^* \nabla^2 \left(\left(2 + \frac{H_0 - U}{mc^2} \right)^{-1} \psi \right) d\tau + \int \psi^* U \psi d\tau \quad -(11)$$

where

$$U = -\frac{e^2}{4\pi F_0 r} \quad -(12)$$

Now use:

$$\frac{1}{2 + \frac{H_0 - U}{mc^2}} = \frac{1}{2} \left(\frac{1}{1 + \frac{H_0 - U}{2mc^2}} \right) \sim \frac{1}{2} \left(1 - \frac{H_0 - U}{2mc^2} \right) \quad -(13)$$

$$\text{if } H_0 - U \ll 2mc^2 \quad -(14)$$

So:

$$\begin{aligned} \langle H_0 \rangle &= -\frac{e^2}{2m} \int \psi^* \nabla^2 \psi d\tau + \int \psi^* U \psi d\tau \\ &\quad + \frac{e^2}{4mc^2} \int \psi^* \nabla^2 ((H_0 - U) \psi) d\tau \end{aligned} \quad -(15)$$

i.e.

$$\langle H_0 \rangle = -\frac{me^4}{32\pi^2\epsilon_0^2\hbar^2n^2} + \frac{\hbar^2}{4m^2c^2} \int \psi^* \nabla^2 ((H_0 - U)\psi) d\tau \quad -(16)$$

There is a shift in the energy levels of the atom which is different for each n .

Now use the fact that H_0 is a constant of motion. Therefore:

$$\langle H_0 \rangle = -\frac{me^4}{32\pi^2\epsilon_0^2\hbar^2n^2} + \frac{\hbar^2 H_0}{4m^2c^2} \int \psi^* \nabla^2 \psi d\tau - \frac{\hbar^2}{4m^2c^2} \int \psi^* \nabla^2 (U\psi) d\tau \quad -(17)$$

In the first approximation eq (16) can be used for H_0 on the right hand side of eq. (17), so:

$$\langle H_0 \rangle = -\frac{me^4}{32\pi^2\epsilon_0^2\hbar^2n^2} \left(1 + \frac{\hbar^2}{4m^2c^2} \int \psi^* \nabla^2 \psi d\tau \right) - \frac{\hbar^2}{4m^2c^2} \int \psi^* \nabla^2 (U\psi) d\tau \quad -(18)$$

→ The shift in the energy levels can be worked out using the hydrogenic wavefunctions for ψ . Note that:

$$\begin{aligned}\nabla^2(\bar{u}\psi) &= \underline{\nabla} \cdot \underline{\nabla}(\bar{u}\psi) \\ &= \underline{\nabla} \cdot (\psi \underline{\nabla} \bar{u} + \bar{u} \underline{\nabla} \psi)\end{aligned}\quad -(19)$$

using the Leibnitz theorem. Similarly:

$$\begin{aligned}\underline{\nabla} \cdot (\psi \underline{\nabla} \bar{u} + \bar{u} \underline{\nabla} \psi) &= \underline{\nabla} \psi \cdot \underline{\nabla} \bar{u} + \psi \nabla^2 \bar{u} \\ &\quad + \underline{\nabla} \bar{u} \cdot \underline{\nabla} \psi + \bar{u} \nabla^2 \psi\end{aligned}\quad -(20)$$

$$= \psi \nabla^2 \bar{u} + \bar{u} \nabla^2 \psi + 2 \underline{\nabla} \psi \cdot \underline{\nabla} \bar{u}.$$

So:

$$\begin{aligned}\langle H_0 \rangle &= -\frac{me^4}{32\pi^2 \epsilon_0^2 \hbar^2 n^2} \left(1 + \frac{\hbar^2}{4m^2 c^2} \int \psi^* \nabla^2 \psi d\tau \right) \\ &\quad - \frac{\hbar^2}{4m^2 c^2} \left[\int \psi^* \nabla^2 \bar{u} \psi d\tau + \int \psi^* \bar{u} \nabla^2 \psi d\tau \right. \\ &\quad \left. + 2 \int \psi^* \underline{\nabla} \bar{u} \cdot \underline{\nabla} \psi d\tau \right]\end{aligned}\quad -(21)$$

So there are several types of shift.

As in HF Theory in the non-relativistic limit:

$$\langle H_0 \rangle = -\frac{\hbar^2}{2m} \int \psi^* \nabla^2 \psi d\tau + \int \psi^* U \psi d\tau$$

$$= \frac{e^2}{8\pi \epsilon_0 r} - \frac{e^2}{4\pi \epsilon_0 r} = -\frac{e^2}{8\pi \epsilon_0 r} \quad -(22)$$

where

$$r = \frac{4\pi \epsilon_0 n^2 \hbar^2}{me^2} \quad -(23)$$

r is Bohr radius.

Comparison with Dirac Approximation

In the Dirac approximation eq. (1) is written as:

$$H_0 = H - mc^2 = \frac{p^2 c^2}{E - U + mc^2} = \frac{p^2 c^2}{E + mc^2} \quad -(24)$$

Dirac assumed that:

$$H \sim E \sim mc^2 \quad -(25)$$

i.e

$$U \ll E \sim mc^2 \quad -(26)$$

$$\text{So } H_0 \sim \frac{p^2 c^2}{2mc^2 - U} + U \quad -(27)$$

$$= \frac{p^2}{2m} \left(1 - \frac{U}{2mc^2} \right)^{-1} + U$$

$$\sim \frac{p^2}{2m} \left(1 + \frac{U}{2mc^2} \right) + U$$

From this approximation:

$$\begin{aligned}\langle H_0 \rangle &= -\frac{\hbar^2}{2m} \int \psi^* \nabla^2 \psi d\tau + \int \psi^* \bar{U} \psi d\tau \\ &\quad - \frac{\hbar^2}{4m^2 c^2} \int \psi^* \nabla^2 (\bar{U} \psi) \quad -(28) \\ &= -\frac{me^4}{32\pi^3 \epsilon_0^3 \hbar^3 n^3} - \frac{\hbar^2}{4m^2 c^2} \int \psi^* \nabla^2 (\bar{U} \psi)\end{aligned}$$

Comparison of eqns (21) and (28) shows that the Dirac approximation misses the term:

$$\boxed{\begin{aligned}\langle H_0 \rangle_1 &= -\frac{me^4}{32\pi^3 \epsilon_0^3 \hbar^3 n^3} \left(\frac{\hbar^2}{4m^2 c^2} \int \psi^* \nabla^2 \psi d\tau \right) \\ &= -\frac{e^4}{128\pi^3 \epsilon_0^3 m c^2 n^3} \int \psi^* \nabla^2 \psi d\tau \quad -(29)\end{aligned}}$$

and this could be evaluated by computer algebra and looked for spectroscopically.